

論文 / 著書情報
Article / Book Information

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Category(English)	Doctoral Thesis
種別(和文)	論文要旨
Type(English)	Summary

(博士課程)
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論文要旨

THESIS SUMMARY

専攻： 有機・高分子物質 専攻
Department of
学生氏名： 金 南均
Student's Name (Kim Nam-Kyun)

申請学位(専攻分野)： 博士 (工学)
Academic Degree Requested Doctor of
指導教員(主)：
Academic Supervisor(main)
指導教員(副)：
Academic Supervisor(sub)

要旨 (英文 800 語程度)

Thesis Summary (approx.800 English Words)

Title of this thesis is "Development of Effective Method for Catalytic Fixation of Carbon Dioxide into Polymers" with 5 chapters. Herein, efficient fixation of carbon dioxide (CO₂) into materials including polymers is described. CO₂-fixation reaction with propargylamine derivatives and its polymers were efficiently transformed to corresponding poly(oxazolidinone)s using various metal- or organo- catalysts.

In chapter 1, "Introduction", the author described the various strategy for CO₂-fixation reaction into the valuable materials and polymers to reveal the purpose and meaning of this thesis.

In chapter 2, "CO₂-fixation reaction of poly(propargylamine)s using Cu-macrocyclic catalyst.", the author demonstrated the CO₂-fixation reaction with propargylamines and its polymers using copper macrocyclic catalyst. At first, propargylamine compounds and the copper macrocyclic catalyst were prepared. Because copper has good interaction with alkyne group and also it is useful transition metal for preparing macrocyclic catalyst. And then CO₂-fixation reaction was performed with propargylamines and polypropargylamine. It was shown that copper catalyst and DABCO played synergistic roles in activating both the alkyne and the amino group of propargylic amines in the reaction. Based on this result, polymer bearing propargylamine moiety was synthesized by the Sonogashira-Hagihara coupling reaction of an AB-type monomer. It was possible to obtain polymers, which has propargylamine moiety in the main chain. A variety of molecular weight polymers were synthesized by controlling reaction temperature and time. It was found that the copper macrocyclic catalyzed polymer transformation reaction of high molecular weight was faster than low molecular weight polymer, although same amount of copper macrocyclic catalyst was used for CO₂-fixation reaction. These results indicate the possibility that the macrocyclic catalyst formed pseudo-rotaxane structure with polymer. Because of this interlocked structure, macrocyclic catalyst continuously react with next propargylamine group. Which is rotaxane-based catalytic system that is consisted with mechanical axle-wheel linkage. The axle component bearing propargylamine was efficiently translated by CO₂-fixation reaction by copper macrocyclic as a wheel component.

In chapter 3, "Base-mediated highly efficient CO₂-fixation to propargylamines and polypropargylamine", the author described the CO₂-fixation reaction with propargyl amines using base-catalysts. Various base-catalysts were used for CO₂ fixation reaction. 1,8-Diazabicyclo[5.4.0]undec-7-ene (DBU) showed good catalytic ability for CO₂-fixation reaction. And also the solvent effect and substituent effect were revealed. It was found that the propargylamines bearing pyridine moieties reacted with CO₂ to poly(oxazolidinone)s in high efficiency under mild conditions, which required only five minutes at room temperature in the presence of DBU under CO₂ atmosphere (0.1 MPa) in DMSO as a solvent. To demonstrate that electronic effect and clarify the reaction mechanism, DFT calculation was performed. This transformation was confirmed by NMR and IR. *E* / *Z* conformation of the double bond of the oxazolidinone moiety was determined by NOE correlation, suggesting the preferential formation of *Z*-isomer under the conditions. The transformation of alkyne group to double bond was clearly indicated by appearance of carbonyl and double bond stretching signal in the IR spectra. This reaction proceeded quantitatively even under atmospheric air condition at room temperature. Moreover, transformed poly(oxazolidinone)s were showed excellent thermal stability (*T*_{d5}: 350 °C).

In chapter 4, "Metal- and solvent-free CO₂-fixation to polypropargylamine", the author described the CO₂-fixation reaction with various polypropargylamines in solvent-free condition. After preparation of polypropargylamines bearing 2-pyridyl group, CO₂-fixation reaction was performed without solvent. It was found that the polypropargylamine reacted with CO₂ to polyoxazolidinones in the presence catalytic amount of DBU under CO₂ atmosphere (0.1 MPa) without a solvent. Additionally, side-chain type polymers were synthesized by via the typical free-radical polymerization using AIBN as an initiator in toluene. Since this polymer system can be modified by copolymerization, catalytic functional group was introduced. Copolymerization of methacrylate bearing propargylamine moiety and propargyl (diethylamino)ethyl methacrylate (DEMA) was prepared. This copolymer containing DEMA showed acceleration of solvent-free CO₂-fixation reaction in the presence catalytic amount of DBU under CO₂ atmosphere (0.1 MPa).

In chapter 5, "Conclusion" summarized the results obtained in this work along with its future prospect.

In summary, to establish the CO₂ incorporation into the polymers, the author designed novel poly(propargylamine)s

bearing propargylamine moiety. Synthesized propargylamines and its polymers are efficiently transformed to poly(oxazolidinone)s by CO₂-fixation reaction. It was found that the copper macrocycle catalyzed polymer transformation reaction system showed pseudo-rotaxane catalyst system. Especially, the propargylamines bearing pyridine moieties transformed to corresponding poly(oxazolidinone)s in high efficiency. In the atmospheric CO₂ pressure and temperature, synthesized poly(propargylamine)s were fixed the CO₂ within few minutes in the presence of DBU. In addition, poly(propargylamine)s were transformed to poly(oxazolidinone)s even open-air condition at room temperature. And also, the solvent free CO₂-fixation reaction were performed. Finally, incorporation of catalytic functional group into the poly(propargylamine) was demonstrated.

備考：論文要旨は、和文 2000 字と英文 300 語を 1 部ずつ提出するか、もしくは英文 800 語を 1 部提出してください。

Note : Thesis Summary should be submitted in either a copy of 2000 Japanese Characters and 300 Words (English) or 1 copy of 800 Words (English).

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